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Ionic Compounds With Polyatomic Ions, Part 3 Quiz: Solutions

$$+ 2 - 1 - 1 = 0$$

$$+ 2 + 2 + 2 - 3 - 3 = 0$$

A. Name the following compounds.

$\text{Cd}_3(\text{PO}_4)_2$ Cadmium Phosphate

$\text{Ba}(\text{BrO}_2)_2$ Barium Bromite

$\text{Ba}(\text{BrO}_4)_2$ Barium Perbromate

$\text{Ra}_3(\text{AsO}_4)_2$ Radium Arsenate

$\text{Cd}(\text{H}_2\text{PO}_4)_2$ Cadmium Dihydrogen Phosphate

B. Write formulas for the following compounds.

Calcium Hydroxide



Beryllium Phosphate



Magnesium Arsenate



Strontium Bromite



Calcium Periodate

