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## Ionic Compounds With Polyatomic Ions, Part 3 Quiz

$$+ 2 - 1 - 1 = 0$$

$$+ 2 + 2 + 2 - 3 - 3 = 0$$

A. Name the following compounds.

$\text{Cd}_3(\text{PO}_4)_2$  \_\_\_\_\_

$\text{BaBrO}_2$  \_\_\_\_\_

$\text{BaBrO}_4$  \_\_\_\_\_

$\text{Ra}_3(\text{AsO}_4)_2$  \_\_\_\_\_

$\text{Cd}(\text{H}_2\text{PO}_4)_2$  \_\_\_\_\_

B. Write formulas for the following compounds.

Calcium Hydroxide \_\_\_\_\_

Beryllium Phosphate \_\_\_\_\_

Magnesium Arsenate \_\_\_\_\_

Strontium Bromite \_\_\_\_\_

Calcium Periodate \_\_\_\_\_