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Ionic Compounds With Polyatomic Ions, Part 3

Get to zero!

$$+ 2 - 1 - 1 = 0$$

$$+ 2 + 2 + 2 - 3 - 3 = 0$$

A.	Ca ⁺²	NO ₃ ⁻¹	NO ₃ ⁻¹	Ca(NO ₃) ₂	Calcium Nitrate		
	Ca ⁺²	NO ₂ ⁻¹	NO ₂ ⁻¹	Ca(NO ₂) ₂	Calcium Nitrite		
	Mg ⁺²	ClO ⁻¹	ClO ⁻¹	Mg(ClO) ₂	Magnesium Hypochlorite		
	Be ⁺²	ClO ₂ ⁻¹	ClO ₂ ⁻¹	Be(ClO ₂) ₂	Beryllium Chlorite		
	Zn ⁺²	ClO ₃ ⁻¹	ClO ₃ ⁻¹	Zn(ClO ₃) ₂	Zinc Chlorate		
B.	Ca ⁺²	Ca ⁺²	Ca ⁺²	PO ₄ ⁻³	PO ₄ ⁻³	Ca ₃ (PO ₄) ₂	Calcium Phosphate
	Ca ⁺²	Ca ⁺²	Ca ⁺²	BO ₃ ⁻³	BO ₃ ⁻³	Ca ₃ (BO ₃) ₂	Calcium Borate
	Mg ⁺²	Mg ⁺²	Mg ⁺²	AsO ₄ ⁻³	AsO ₄ ⁻³	Mg ₃ (AsO ₄) ₂	Magnesium Arsenate
	Be ⁺²	Be ⁺²	Be ⁺²	PO ₄ ⁻³	PO ₄ ⁻³	Be ₃ (PO ₄) ₂	Beryllium Phosphate
	Zn ⁺²	Zn ⁺²	Zn ⁺²	PO ₄ ⁻³	PO ₄ ⁻³	Zn ₃ (PO ₄) ₂	Zinc Phosphate