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## Ionic Compounds With Polyatomic Ions, Part 2

Get to zero!

$$+1 + 1 - 2 = 0$$

$$+1 + 1 + 1 - 3 = 0$$

A.	Na <sup>+1</sup>	Na <sup>+1</sup>	SO <sub>4</sub> <sup>-2</sup>	Na <sub>2</sub> SO <sub>4</sub>	Sodium Sulfate	
	Na <sup>+1</sup>	Na <sup>+1</sup>	SO <sub>3</sub> <sup>-2</sup>	Na <sub>2</sub> SO <sub>3</sub>	Sodium Sulfite	
	K <sup>+1</sup>	K <sup>+1</sup>	CO <sub>3</sub> <sup>-2</sup>	K <sub>2</sub> CO <sub>3</sub>	Potassium Carbonate	
	Li <sup>+1</sup>	Li <sup>+1</sup>	O <sub>2</sub> <sup>-2</sup>	Li <sub>2</sub> O <sub>2</sub>	Lithium Peroxide	
	Ag <sup>+1</sup>	Ag <sup>+1</sup>	CrO <sub>4</sub> <sup>-2</sup>	Ag <sub>2</sub> CrO <sub>4</sub>	Silver Chromate	
B.	Na <sup>+1</sup>	Na <sup>+1</sup>	Na <sup>+1</sup>	PO <sub>4</sub> <sup>-3</sup>	Na <sub>3</sub> PO <sub>4</sub>	Sodium Phosphate
	Na <sup>+1</sup>	Na <sup>+1</sup>	Na <sup>+1</sup>	BO <sub>3</sub> <sup>-3</sup>	Na <sub>3</sub> BO <sub>3</sub>	Sodium Borate
	K <sup>+1</sup>	K <sup>+1</sup>	K <sup>+1</sup>	AsO <sub>4</sub> <sup>-3</sup>	K <sub>3</sub> AsO <sub>4</sub>	Potassium Arsenate
	Li <sup>+1</sup>	Li <sup>+1</sup>	Li <sup>+1</sup>	PO <sub>4</sub> <sup>-3</sup>	Li <sub>3</sub> PO <sub>4</sub>	Lithium Phosphate
	Ag <sup>+1</sup>	Ag <sup>+1</sup>	Ag <sup>+1</sup>	PO <sub>4</sub> <sup>-3</sup>	Ag <sub>3</sub> PO <sub>4</sub>	Silver Phosphate