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Ionic Compounds With Polyatomic Ions, Part 1

Get to zero!

$$+ 1 - 1 = 0$$

$$+ 2 - 2 = 0$$

$$+ 3 - 3 = 0$$

A.	Na ⁺¹	NO ₃ ⁻¹	NaNO ₃	Sodium Nitrate
	Na ⁺¹	NO ₂ ⁻¹	NaNO ₂	Sodium Nitrite
	K ⁺¹	ClO ⁻¹	KClO	Potassium Hypochlorite
	Li ⁺¹	ClO ₂ ⁻¹	LiClO ₂	Lithium Chlorite
	Ag ⁺¹	ClO ₃ ⁻¹	AgClO ₃	Silver Chlorate
B.	Ca ⁺²	SO ₄ ⁻²	CaSO ₄	Calcium Sulfate
	Ca ⁺²	SO ₃ ⁻²	CaSO ₃	Calcium Sulfite
	Mg ⁺²	CO ₃ ⁻²	MgCO ₃	Magnesium Carbonate
	Be ⁺²	O ₂ ⁻²	BeO ₂	Beryllium Peroxide
	Zn ⁺²	CrO ₄ ⁻²	ZnCrO ₄	Zinc Chromate
C.	Al ⁺³	PO ₄ ⁻³	AlPO ₄	Aluminum Phosphate
	Ga ⁺³	BO ₃ ⁻³	GaBO ₃	Gallium Borate
	In ⁺³	AsO ₄ ⁻³	InAsO ₄	Indium Arsenate
	In ⁺³	PO ₄ ⁻³	InPO ₄	Indium Phosphate
	Y ⁺³	PO ₄ ⁻³	YPO ₄	Yttrium Phosphate