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## Ionic Compounds With Roman Numerals, Part 1

Get to zero!

$$+ 1 - 1 = 0$$

$$+ 2 - 2 = 0$$

$$+ 3 - 3 = 0$$

A.	Cu <sup>+1</sup>	Cl <sup>-1</sup>	CuCl	Copper (I) Chloride
	Cu <sup>+1</sup>	Br <sup>-1</sup>	CuBr	Copper (I) Bromide
	Au <sup>+1</sup>	Cl <sup>-1</sup>	AuCl	Gold (I) Chloride
	Tl <sup>+1</sup>	Cl <sup>-1</sup>	TlCl	Thallium (I) Chloride
	Cu <sup>+1</sup>	F <sup>-1</sup>	CuF	Copper (I) Fluoride
B.	Cu <sup>+2</sup>	O <sup>-2</sup>	CuO	Copper (II) Oxide
	Ni <sup>+2</sup>	S <sup>-2</sup>	NiS	Nickel (II) Sulfide
	Co <sup>+2</sup>	Se <sup>-2</sup>	CoSe	Cobalt (II) Selenide
	Fe <sup>+2</sup>	O <sup>-2</sup>	FeO	Iron (II) Oxide
	Sn <sup>+2</sup>	O <sup>-2</sup>	SnO	Tin (II) Oxide
C.	Ni <sup>+3</sup>	N <sup>-3</sup>	NiN	Nickel (III) Nitride
	Co <sup>+3</sup>	P <sup>-3</sup>	CoP	Cobalt (III) Phosphide
	Fe <sup>+3</sup>	P <sup>-3</sup>	FeP	Iron (III) Phosphide
	Cr <sup>+3</sup>	N <sup>-3</sup>	CrN	Chromium (III) Nitride
	V <sup>+3</sup>	N <sup>-3</sup>	VN	Vanadium (III) Nitride