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## Ionic Compounds, Part 3 Quiz

$$+ 2 - 1 - 1 = 0$$

$$+ 2 + 2 + 2 - 3 - 3 = 0$$

A. Name the following compounds.

$\text{Cd}_3\text{P}_2$  \_\_\_\_\_

$\text{BaBr}_2$  \_\_\_\_\_

$\text{BaCl}_2$  \_\_\_\_\_

$\text{Ra}_3\text{N}_2$  \_\_\_\_\_

$\text{CdCl}_2$  \_\_\_\_\_

B. Write formulas for the following compounds.

Calcium Iodide \_\_\_\_\_

Beryllium Phosphide \_\_\_\_\_

Magnesium Nitride \_\_\_\_\_

Strontium Chloride \_\_\_\_\_

Calcium Fluoride \_\_\_\_\_