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Ionic Compounds, Part 3

Get to zero!

$$+ 2 - 1 - 1 = 0$$

$$+ 2 + 2 + 2 - 3 - 3 = 0$$

A.	Ca ⁺²	Cl ⁻¹	Cl ⁻¹	CaCl ₂	Calcium Chloride		
	Ca ⁺²	Br ⁻¹	Br ⁻¹	CaBr ₂	Calcium Bromide		
	Mg ⁺²	Cl ⁻¹	Cl ⁻¹	MgCl ₂	Magnesium Chloride		
	Be ⁺²	Cl ⁻¹	Cl ⁻¹	BeCl ₂	Beryllium Chloride		
	Zn ⁺²	Cl ⁻¹	Cl ⁻¹	ZnCl ₂	Zinc Chloride		
B.	Ca ⁺²	Ca ⁺²	Ca ⁺²	N ⁻³	N ⁻³	Ca ₃ N ₂	Calcium Nitride
	Ca ⁺²	Ca ⁺²	Ca ⁺²	P ⁻³	P ⁻³	Ca ₃ P ₂	Calcium Phosphide
	Mg ⁺²	Mg ⁺²	Mg ⁺²	P ⁻³	P ⁻³	Mg ₃ P ₂	Magnesium Phosphide
	Be ⁺²	Be ⁺²	Be ⁺²	N ⁻³	N ⁻³	Be ₃ N ₂	Beryllium Nitride
	Zn ⁺²	Zn ⁺²	Zn ⁺²	N ⁻³	N ⁻³	Zn ₃ N ₂	Zinc Nitride