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Ionic Compounds, Part 2

Get to zero!

$$+1 + 1 - 2 = 0$$

$$+1 + 1 + 1 - 3 = 0$$

A.	Na ⁺¹	Na ⁺¹	O ⁻²	Na ₂ O	Sodium Oxide	
	Na ⁺¹	Na ⁺¹	S ⁻²	Na ₂ S	Sodium Sulfide	
	K ⁺¹	K ⁺¹	Se ⁻²	K ₂ Se	Potassium Selenide	
	Li ⁺¹	Li ⁺¹	O ⁻²	Li ₂ O	Lithium Oxide	
	Ag ⁺¹	Ag ⁺¹	O ⁻²	Ag ₂ O	Silver Oxide	
B.	Na ⁺¹	Na ⁺¹	Na ⁺¹	N ⁻³	Na ₃ N	Sodium Nitride
	Na ⁺¹	Na ⁺¹	Na ⁺¹	P ⁻³	Na ₃ P	Sodium Phosphide
	K ⁺¹	K ⁺¹	K ⁺¹	P ⁻³	K ₃ P	Potassium Phosphide
	Li ⁺¹	Li ⁺¹	Li ⁺¹	N ⁻³	Li ₃ N	Lithium Nitride
	Ag ⁺¹	Ag ⁺¹	Ag ⁺¹	N ⁻³	Ag ₃ N	Silver Nitride