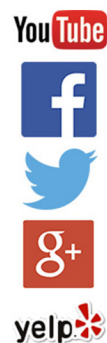




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Ionic Compounds, Part 1

Get to zero!

$$+ 1 - 1 = 0$$

$$+ 2 - 2 = 0$$

$$+ 3 - 3 = 0$$

A.	Na ⁺¹	Cl ⁻¹	NaCl	Sodium Chloride
	Na ⁺¹	Br ⁻¹	NaBr	Sodium Bromide
	K ⁺¹	Cl ⁻¹	KCl	Potassium Chloride
	Li ⁺¹	Cl ⁻¹	LiCl	Lithium Chloride
	Ag ⁺¹	Cl ⁻¹	AgCl	Silver Chloride
B.	Ca ⁺²	O ⁻²	CaO	Calcium Oxide
	Ca ⁺²	S ⁻²	CaS	Calcium Sulfide
	Mg ⁺²	Se ⁻²	MgSe	Magnesium Selenide
	Be ⁺²	O ⁻²	BeO	Beryllium Oxide
	Zn ⁺²	O ⁻²	ZnO	Zinc Oxide
C.	Al ⁺³	N ⁻³	AlN	Aluminum Nitride
	Ga ⁺³	P ⁻³	GaP	Gallium Phosphide
	In ⁺³	P ⁻³	InP	Indium Phosphide
	In ⁺³	N ⁻³	InN	Indium Nitride
	Y ⁺³	N ⁻³	YN	Yttrium Nitride