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Polyatomic Ions

+1

Ammonium - NH_4^{+1}

+2

Mercury (I) - Hg_2^{+2}

Mercury (II) - Hg^{+2}

-1

Acetate - $\text{C}_2\text{H}_3\text{O}_2^{-1}$

Bicarbonate (Hydrogen Carbonate) - HCO_3^{-1}

Bisulfate (Hydrogen Sulfate) - HSO_4^{-1}

Bromate - BrO_3^{-1}

Bromite - BrO_2^{-1}

Chlorate - ClO_3^{-1}

Chlorite - ClO_2^{-1}

Cyanate - OCN^{-1}

Cyanide - CN^{-1}

Dihydrogen Phosphate - $\text{H}_2\text{PO}_4^{-1}$

Hydroxide - OH^{-1}

Hypobromite - BrO^{-1}

Hypochlorite - ClO^{-1}

Hypoiodite - IO^{-1}

Iodate - IO_3^{-1}

Iodite - IO_2^{-1}

Nitrate - NO_3^{-1}

Nitrite - NO_2^{-1}

Perbromate - BrO_4^{-1}

Perchlorate - ClO_4^{-1}

Periodate - IO_4^{-1}

Permanganate - MnO_4^{-1}

Thiocyanate - SCN^{-1}

-2

Carbonate - CO_3^{-2}

Chromate - CrO_4^{-2}

Dichromate - $\text{Cr}_2\text{O}_7^{-2}$

Hydrogen Phosphate - HPO_4^{-2}

Oxalate - $\text{C}_2\text{O}_4^{-2}$

Peroxide - O_2^{-2}

Selenate - SeO_4^{-2}

Sulfate - SO_4^{-2}

Sulfite - SO_3^{-2}

Thiosulfate - $\text{S}_2\text{O}_3^{-2}$

-3

Arsenate - AsO_4^{-3}

Borate - BO_3^{-3}

Phosphate - PO_4^{-3}